

Bisphosphonates: An Emerging Trend in Dentistry

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ABSTRACT

Aim: To discuss the basic mechanism alongwith the clinical implications of Bisphosphonates (BPs) in dentistry.

Summary: Bisphosphonates (BPs) are a class of drugs that prevent the loss of bone mass, used to treat osteoporosis and similar diseases. These are compounds used in the treatment of many skeletal disorders such as bone metastases, osteoporosis, Paget's disease etc. and promote the processes of inflammation and destruction by decreasing bone remodeling and exerting antiangiogenic and apoptotic effects. Their clinical application has recently increased in dentistry. BPs have paradoxical effects in the oral cavity, having potential beneficial effects, whilst also increasing the risk of osteo necrosis of jaw. Implant patient who has been taking an oral bisphosphonate for osteoporosis is at the possible risk of developing osteo necrosis of jaw after implant placement. Its various harmful and beneficial effects have been discussed in this paper.

Keywords: Alendronate, Bone Metabolism, Diphosphonate, Bisphosphonate-associated Osteonecrosis of the jaw (BON).

INTRODUCTION

Bisphosphonates (also called diphosphonates) are a class of drugs that prevent the loss of bone mass, and used in the treatment of many skeletal disorders such as bone metastases, osteoporosis, Paget's disease etc.¹ They are called bisphosphonates (BPs) because they have two phosphonate (PO_3) groups. These are structurally similar to natural pyrophosphate (PP), which is a normal product of human metabolism that has a calcium chelating property and is seen in serum and urine (Fig. 1).²

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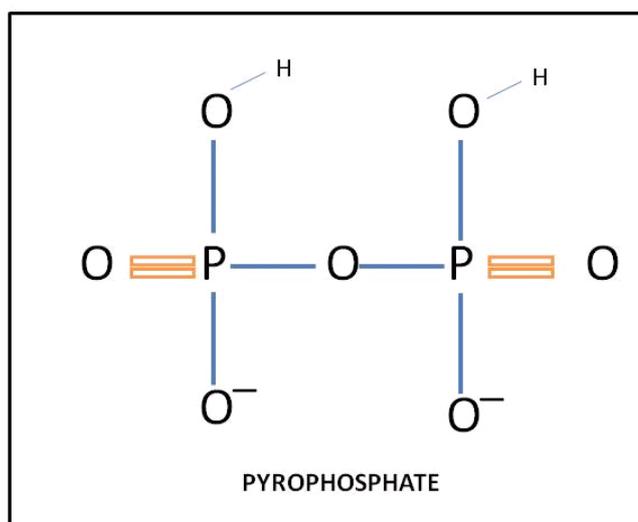


Figure 1: Pyrophosphate

All the geminal BPs and PPs have a fundamental 'planar W' (O-P-C-P-O) configuration that provides bidentate binding to calcium with methylidene BP and the stronger tridentate binding via the hydroxyl group of hydroxymethylidene BP (Fig. 2).³ PP, in contrast to its analogs that are hydrolytically stable, is an unstable molecule in vivo, and its labile P-O-P bond undergoes rapid hydrolysis by pyrophosphatase and alkaline phosphatase activity.⁴

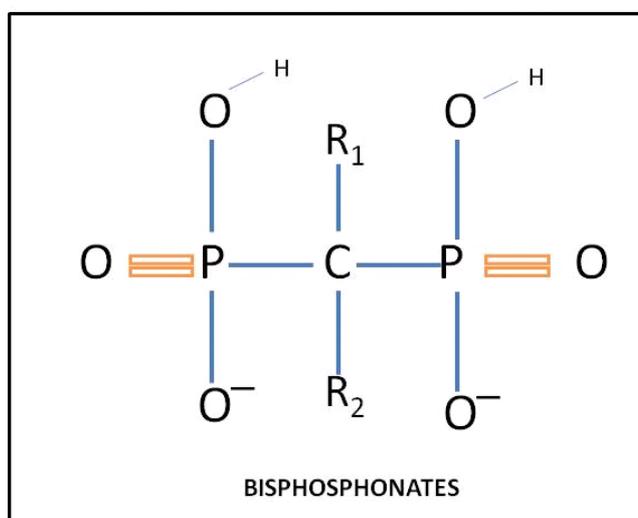


Figure 2: Bisphosphonate

These drugs have a high attraction for hydroxyapatite and rapidly included into all parts of the skeleton.⁵ Also, in addition to decreasing hypercalcemia of malignancy, they are used as inhibitors of osteoclastic activity to alleviate bone pain that results from the release of biochemical mediators in metastatic bone disease.⁶ Despite the beneficial effects, drug-induced osteonecrosis of jaw is an important complication in patients treated by BPs on long-term basis.⁷ Because of its adverse effects, increasing trends of its application in dentistry locally or systemically and its interaction with various dental treatments should be critically reviewed. Present review is intended to discuss the basic mechanism alongwith the clinical implications of BPs in dentistry.

METHODS

We searched the dental literature with Medline/PubMed with an emphasis on peer-reviewed dental journals. MeSH (Medical Subject Headings) terms used were “bisphosphonate” and “dentistry”. Pertinent articles on the topic and abstracts of relevant papers were scrutinised thoroughly. Relevant literature for “bisphosphonate” in common textbooks on pharmacology, periodontology, oral implantology, oral and maxillofacial surgery; bibliographies of papers and review articles together with appropriate peer reviewed journals were also analyzed for additional information.

LITERATURE REVIEW

In 1897, Von Baeyer and Hoffmann⁸ reported the synthesis of the first geminal (central) bisphosphonates, 1-hydroxy-1, 1-ethylidene bisphosphonate disodium salt, but marketable usage for these compounds took place only after their reported use for detergent solutions as complexing agents for calcium and magnesium by Blazer and Worms.³

Scientists of Procter and Gamble (P & G) were investigating the calcium chelating agents, which were found to be effective in removing calculus, and chelated enamel calcium causing surface etch damage.⁹ Henkel Corporation of Germany, introduced a potential builder for detergency, 1-hydroxy-1, 1-ethylidene bisphosphonate disodium salt (EHDP, etidronate) that could chelate calcium and magnesium in the water blocking re-deposition of soil in the wash, to Procter & Gamble, and this extremely effective chelator of calcium found to cause no damage to the highly polished dental enamel surface.³

BPs were originally used in industry, mainly as corrosion inhibitors or as complexing agents in the textile, fertilizer, and oil industries.¹⁰ First human use of a geminal bisphosphonate, etidronate, was reported by Bassett *et al.*¹⁰ in 1969 for the treatment of Myositis Ossificans Progressiva (MPO) and Smith *et al.*¹¹ were the first to report the evidence of the effectiveness of the bisphosphonates for the treatment of Paget's disease of bone. Since then, bisphosphonates have

been approved for the treatment of Paget's disease of bone, hypercalcemia of malignancy using intravenous form, multiple myeloma, or detecting soft tissue calcifications, and for the prevention and treatment of postmenopausal osteoporosis as well as heterotopic ossification due to total hip replacement or spinal cord injury.³

Pharmacological Properties of Bisphosphonates

In bisphosphonates, two phosphonate groups are attached to a geminal (central) carbon that replaces the oxygen in pyrophosphate,¹² which is used as an ingredient in toothpaste to control calculus formation.¹³ Three dimensional structure of this pyrophosphate analog, is capable of chelating divalent cations (Ca^{2+}), thus having a strong affinity for bone surfaces undergoing remodeling.¹² Bisphosphonates forms an ultra-thin layer by chemisorption to the calcium of hydroxyapatite in bones and teeth.³ Concentration gradient of bisphosphonate in the extracellular fluid is high, but due to the stronger adsorption of the bisphosphonate on the inorganic HA than on the organic osteoid it is approaching zero (Quantitative Adsorption Ratio; HA: Osteoid=40:1).^{3,14} Highly negatively charged, bisphosphonates are membrane impermeable and are incorporated into the bone matrix by fluid-phase endocytosis.

Working profile of each bisphosphonate is determined by its unique side chain. The potency and side effects of the compound are changed by the substitution of different side chains for hydrogen at locations R1 and R2. Amongst modified side chains, hydroxyl group at R1 enhances the binding to bone, whereas the cellular effects of bisphosphonates and their relative efficacies as inhibitors of bone resorption may be regulated by R2 structure and 3-dimensional structure.⁶

The bisphosphonates metabolism of bone modulation occurs at 3 levels.⁶ At *tissue* level there is decrease in bone turnover due to decrease in bone resorption and decrease in number of new bone (multicellular units). Thereby, a positive bone balance is maintained in the body. At *cellular* level there is; decrease in osteoclast recruitment, increased osteoclast apoptosis, decreased osteoclast adhesion, decrease depth of resorption site, decrease in release of cytokines by the macrophages, and increase in osteoblasts differentiation and number. At *molecular* level there is inhibition of Mevalonate pathway which is an important cellular metabolic pathway present in all higher eukaryotes and many bacteria. It is important for the production of dimethylallyl pyrophosphate (DMAPP) and isopentenyl pyrophosphate (IPP), which serve as the basis for the biosynthesis of molecules used in processes as diverse as terpenoid synthesis, protein prenylation, cell membrane maintenance, hormones, protein anchoring, and N-glycosylation. It is also a part of steroid biosynthesis and can result in disturbed cell action and induction of apoptosis, and decreased post

translational prenylation of Guanosine triphosphate (GTP)-binding proteins.⁶

Different mode of actions of BPs has been suggested includes: Bisphosphonate-mediated inhibition of the development of osteoclasts, induction of osteoclastic apoptosis,¹⁵ reduction of activity,¹⁶ prevention of the development of osteoclasts from hematopoietic precursors;¹⁷ and stimulation of production of an osteoclast inhibitory factor.¹⁸ It has also been shown that the bisphosphonate alendronate caused a rise in intracellular calcium levels in an osteoclast-like cell line.¹⁹ This finding is of great interest since it could suggest the presence of a receptor for bisphosphonates on osteoclasts. There are various brands of bisphosphonates available in market with varying degree of potency for inhibition of osteoclastic activity. Pyrophosphate drugs are potent inhibitors of osteoclast activity that also have anti-angiogenic effects. The drugs have a high affinity for hydroxyapatite and are rapidly incorporated into all parts of the skeleton and have a very long half-life. Relative potencies of the agents depend on their formulation.

Bisphosphonates have been divided under various generations; the first generation of bisphosphonates include etidronate which was initially used to inhibit ectopic calcification and later was used as to prevent resorption, the second generation bisphosphonates like Clodronate were used as drugs to prevent bone resorption rather than preventing mineralization to occur, and the third generation drugs had greater potency than first and second generation bisphosphonates (Table 1).^{12,20} Amongst various BPs, nitrogen containing bisphosphonates (nBPs) that inhibit farnesyl diphosphonate (FPP) synthetase in the mevalonate pathway, exert intense pharmacodynamic effects long after their blood levels reach zero, and approximately half of any nBP doses reaches the skeleton with an early half life of 10 days and a terminal half-life of about 10 years. FPP synthetase is responsible for isoprenylation of small GTPase as well as geranylgeranylation of Rho and Rac, which promote an array of activities in the osteoclasts that control bone resorption. At low concentrations, the fundamental activities which are related to bone mineral dissolution and collagen degradation, that involve the cytoskeleton, vesicular trafficking, and membrane ruffling are inhibited. At somewhat high concentrations, osteoclast differentiation is inhibited and at concentration of 100 μ M, osteoclast apoptosis is induced by Rho and Rac.²¹

Being highly polar compounds, BPs are poorly absorbed after oral ingestion with the bioavailability of less than 5%.^{22,23} In order to reduce the interference with the food, timing of meal is important, and to increase the amount of BPs introduced to bone intravenous administration of the drug can also be advised.²² Once in the blood stream, almost the entire drug dose is either eliminated in urine or absorbed by the bone,²⁴

Table 1: Generations of Bisphosphonates

Drugs	Properties
1 st Generation e.g. Etidronate, Medronate, Clodronate, Tiludronate.	Minimally modified side chains (R1 R2) or contain a chlorophenyl group. Can cause bone demineralisation. Apoptosis account for antiresorptive effect. Metabolised into a nonhydrolyzable ATP analog that accumulates within oteoclasts and induces apoptosis. Least potent.
2 nd Generation e.g. Alendronate, Pamidronate, Ibandronate.	Aminobisphosphonates, contain nitrogen group (amino terminal) in the side chain. Primarily inhibits bone resorption rather than preventing mineralization. Antiresorptive activity involves inhibition of components of the cholesterol biosynthesis pathway. (G & G) Directly inhibit multiple steps in the pathway from mevalonate to cholesterol and isoprenoid lipids (as geranylgeranyl diphosphonate) that are required for the prenylation of proteins that are important for osteoclast function. They are 10-100 times more potent than 1 st generation.
3 rd Generation eg. Risedronate, Zoledronate	Contain nitrogen atom within a heterocyclic ring. These are upto 10,000 times more potent than 1 st generation.

and in owing to its negative charge and chemical structure, it can be retained as long as ten years.²⁵ During bone remodelling, BPs are released into the acidic environment of the resorption lacunae where they impede osteoclast bone resorptive action.²⁶ Osteoclasts take up bisphosphonates from resorption lacunae in the bone matrix, and the bisphosphonates then trigger the osteoclasts to undergo apoptosis.²⁷

Bone is normally able to adapt to functional changes through specialized cells, represented by osteoblasts, osteoclasts, and lining cells. This drug is potent inhibitors of osteoclast activity that also have anti-angiogenic effects.⁵ Since bone damage is caused by increased numbers and activity of these osteoclast bone cells, bisphosphonates reduce new bone damage and allow an opportunity for bone healing to occur. Bisphosphonates therefore have several beneficial effects, including:²⁸ Preventing further bone damage; Reducing bone pain and the need for painkillers; Correcting and preventing hypercalcemia (higher than normal levels of calcium in the blood); Reducing the need for radiotherapy; Reducing pathologic fractures due to myeloma (i.e., fracture at a site where myeloma has weakened the bone); Improving quality

of life; and Improving the chances of healing and recovery of strength of the bone.²⁸

Adverse Effects of Bisphosphonates

The various toxic effects of BPs can be divided into: *Common* that include renal toxicity, acute-phase reactions, gastrointestinal toxicity; *Rare* as hypocalcemia (symptomatic), ocular complications (retinitis, uveitis, scleritis), asthma (aspirin-sensitive), erythema, phlebitis, altered taste, central nervous system side effects; and *Emerging* as osteonecrosis of the jaw.²⁹ Besides above mentioned effects, nBPs may inhibit angiogenesis by reducing circulating levels VEGF (vascular endothelial growth factor), and is also found to be associated with accumulation of microcracks at or somewhat above osteoporosis doses in dogs.²¹

Numerous studies and reports have revealed the renal toxic action of the bisphosphonates.³⁰⁻³⁴ Varying renal histopathology of acute tubular necrosis associated with individual bisphosphonates may depend on the degree to which the drugs accumulate in the renal parenchyma, which, in turn, depends on the drugs' pharmacokinetics and pharmacodynamics. Renal toxicity for intravenous bisphosphonates determined by the dosage, frequency and speed of infusion, but, so far renal complications following therapeutic oral bisphosphonates have not been reported.²⁹

Acute-phase reactions, characterised by number of flu-like signs and symptoms, particularly subfebrile temperature (38°C/100.4°F), leukocytosis, exhaustion, and muscle and bone pain, occur only with intravenously administered aminobisphosphonates (zoledronic acid, ibandronate, and pamidronate), usually after the first infusion.³⁵⁻³⁷ Transient increase in pyrogenic cytokines (IL-6, TNF-alpha) may cause acute-phase reactions.³⁸ Although it is not life-threatening, these reactions may sometimes causing treatment withdrawal and symptoms generally resolve within 48 hours that respond well to nonsteroidal anti-inflammatory drugs and antipyretics.²⁹

Bisphosphonate-induced adverse effects in the Gastrointestinal (GI) tract are naturally seen only after oral therapy and lifethreatening complications are extremely rare, however, if adverse GI reactions persist, then intravenous route may be preferred.²⁹ All levels of the GI tract can be affected, from the lower esophagus to the colon, although ulceration in the esophagus, stomach, and duodenum can occur, mucositis, flatulence, and diarrhea are more common.³⁹⁻⁴¹

Osteonecrosis of the jaw or Bisphosphonate-Associated Osteonecrosis (BON): The term osteonecrosis (Osteo = bone + necrosis = cell or tissue death) simply means death of bone

tissue; bone die in any part of the body if its blood supply is cut off and the cells can't get oxygen or food.¹ Marx,⁴² Milgiorati,⁴³ and Ruggiero *et al.*⁴⁴ were amongst the pioneers to describe BON in association with bisphosphonate treatment. BON has similarities with the almost disappeared condition, and presently only reported in Chinese firework factory workers, "phossy jaw" first observed in the 19th century in individuals exposed to white (yellow) phosphorus.^{29,45,46}

Clinical presentation of BON includes soft tissue swelling and exposed, necrotic bone that has persisted for more than eight weeks^{47,48} in patients who have taken or be currently using BPs, while there is no other perplexing conditions (eg. radiotherapy of jaw, alcoholism etc).²⁶ The vast majority of BON cases occur in cancer patients who have received high-potency aminobisphosphonates given intravenously to decrease the osteolytic effects of multiple myeloma or malignancies that have metastasized to bone.⁵

With female sex predilection (3:2), BON is more commonly identified in mandible (posterior lingual side near mylohyoid ridge) alone followed by maxilla and then in both maxilla and mandible.^{26,50} Not all patients taking BPs develop BON and frequency of occurrence of BON is between 1% and 12%,⁴⁹⁻⁵¹ and less than 1% in patients on oral BPs.⁴⁹ There are putative risk factors that may predispose the BON condition which include: age; tobacco use and alcoholism; route of administration (intravenous); Cumulative dosage and potency of bisphosphonates; medical conditions like diabetes, coagulopathy, blood dyscrasias, malignancy; dentoalveolar surgery, intraoral trauma, periodontal disease, poor oral hygiene, poorly fitting dentures, prominent mandibular lingual ridge, mylohyoid ridge and tori; cancer and anticancer therapy, glucocorticoid and estrogen therapy; and low mineral density of skeletal systems.²⁶ All aminobisphosphonates may cause osteonecrosis of jaw with long-term use.⁵²

Usually jaw necrosis is characterized clinically by chronically exposed jawbone and including suppuration and sequestration. These lesions do not respond to treatment or intensive antibiotic therapy and repeated jaw surgery.^{26,53} Osteonecrosis of jaw may develop due to dentogenic portals of pathogen entry (eg. periodontal disease and periapical granuloma) or to prosthetic pressure points, leading to endosteal infiltration of the jawbone. Bisphosphonates promote the processes of inflammation and destruction by decreasing bone remodelling and exerting antiangiogenic and apoptotic effects.⁵⁴ The presence of persistent alveolar sockets after teeth extractions is an important radiographic sign in the jaws associated with BON.⁵⁵ Individual with BON often have difficult chewing, speaking and swallowing and severity is comparable to osteonecrosis after radiotherapy to the head and neck.²⁹

Implications of Bisphosphonates in Dentistry

Many studies have shown the potential of topical BPs application to enhance osseointegration of dental implants.^{46,56-58} Its application on dental implants, with or without calcium phosphate layer promoted implant-bone contact^{57,58} and increased the amount of bone peripheral to implants in dogs.⁵⁹ Zuffetti *et al.*²⁰ reported that bisphosphonate-treated implant showed more contact with newly formed bone than the control implant. However, despite these potential benefits, it may contribute the development of BON. Implant patient who has been taking an oral bisphosphonate for osteoporosis is the possible risk of developing osteonecrosis of jaw after implant placement. Oral BPs have been reported to be associated with implant failure^{59,60} and Cheng *et al.*⁶¹ reported this risk to be 0.88% of the patients receiving oral BPs. The risk is real as bisphosphonates bind to hydroxyapatite and have a very long half-life; it is likely that the length of time a patient has been taking oral bisphosphonates is important in determining the level of risk. Since oral bisphosphonates slowly accumulate in bone with time, an osteoporosis patient who has been taking the drug for one year is at a lower risk of developing osteonecrosis of jaw or implant failure than someone who has been on the drug for many years. In general, it is not recommended that implants be placed in patients who have been on the drug for more than 3 years. Prolonged use of bisphosphonates is a contraindication to implant placement.⁶²

The use of systemic BPs has shown to reduce alveolar bone loss in large animal models which were experimentally induced or had naturally occurring periodontitis without significantly affecting clinical periodontal parameters.⁶³ Controlled clinical trials in humans showed the efficacy of alendronate in reducing the alveolar bone loss relative to placebo and in one study the effect was seen in subgroup with low mandibular bone density.⁶⁴⁻⁶⁷ Kaynak⁶⁸ did a study on male rats and stated when systemic administration of 0.5 mg/kg was effective enough for preventing alveolar bone loss and in modulating tissue factors. These findings indicate that alendronate is a valuable therapeutic medicine which can be used for the treatment of periodontal disease either alone or in combination with regenerative components like anti-inflammatory drug, bone graft, material and guided tissue regeneration or even with dental implants.⁶⁸ Lane *et al.*⁶⁹ suggested that BPs treatment improves the clinical outcome of non-surgical periodontal therapy and may be an appropriate adjunctive treatment to severe periodontal bone loss. A newly developed BP, TRK-530 (disodium dihydrogen [4-(methylthio) phenylthio] methanbisphosphonate), has recently been shown to have anti-inflammatory, anti-bone-resorbing activity as well as dose dependent local anticalculogenic action.⁷⁰ Houshmand *et al.*⁷¹ in an in-vivo

study evaluated boosting effect of nBP (Pamidronate disodium) on osteoconductive material and revealed improved bone healing process as a result of reducing the number of osteoclasts and increasing the amount of trabeculae in regenerated tissue.

BPs having high affinity for calcium phosphate crystals and that inhibit osteoclast activity, also appear to inhibit MMP activity through a mechanism that involves chelation of cations,^{72,73} however, minimal effects were demonstrated on clinical parameters.⁷⁴ Topically administered bisphosphonates have been reported to reduce the resorption of root associated with orthodontic tooth movement and alveolar bone resorption following periodontal surgery.^{75,76} BPs also reported to modulate cementoblast behavior in an in-vitro study through the regulation of gene expression, and thus has the potential for cementum formation and mineralisation modifiers.¹³ BPs however, have paradoxical effects in the oral cavity, having potential beneficial effects on periodontal disease, whilst also increasing the risk of osteonecrosis of jaw and hence, the ethical issues inherent in undertaking such trials are likely to limit further trials in humans.⁴⁶ Future randomized, controlled, longitudinal studies that evaluate therapies for the treatment of osteoporosis should also examine the effectiveness of the treatment on periodontal disease parameters.⁷⁴

For the maintenance of alveolar bone after tooth extraction, submerged hydroxyapatite implants have been attempted. Denssen *et al.*⁷⁷ suggested that normal osteoconduction and repair occurred in alveolar bone around the highly bisphosphonate-complexed hydroxyapatite implant. However, the long term effect of the presence of BPs in the healed alveolar bone remained unclear.

Bisphosphonates are very effective in managing osteogenesis imperfecta in children, but no cases of osteonecrosis of jaw have been reported, suggesting extractions are not contraindicated in these children. Young age might possibly be a protective factor for prevention of osteonecrosis of jaw.⁷⁸⁻⁸⁰ Caution is advised with invasive diode laser therapy, miniscrew skeletal anchorage devices, mucosal trauma from retainers, orthognathic surgery and tooth extraction. It has also been proposed that patients discontinue their bisphosphonate therapy for a period of time prior to orthodontic treatment.⁸¹ This concept would require further investigation as the half life of bisphosphonates is approximately 10 years. It has been seen in an experimental study involving local administration of bisphosphonates in rats that there is a positive role for topical bisphosphonates in orthodontic treatment, inhibiting undesirable movement of anchor teeth and inhibiting post-treatment relapse in a dose dependent manner.^{82,83} Extrapolation of these animal models to humans requires special care as there is a risk of osteonecrosis of jaw. Endodontic treatment is the preferred

treatment over extraction to minimize osteo necrosis of jaw risk. Due to bisphosphonate suppression of bone turnover, the periapical rarefaction may not decrease in size in a manner comparable to patients with normal bone turnover. Strathy *et al.*⁸⁴ described the development of osteo necrosis of jaw in the region of teeth requiring endodontic treatment in patients on long-term BPs.²⁹ Endodontic treatment has not been considered as a direct risk for osteonecrosis of jaw, but care has to be taken to minimize trauma due to rubber dam placement and filling beyond the apex, apicoectomy is contraindicated as any type of injury in patients on long term bisphosphonate therapy is a positive risk factor for BON. Orthodontic elastics should be used to promote atraumatic extraction. Extractions should always be performed as atraumatically as possible with direct closure of the socket by suturing and antibiotic prophylaxis considered especially in immunocompromised patients.⁴⁶ Similarly prosthodontic appliances should be adjusted for fit to avoid mucosal irritation. Levin *et al.*⁸⁵ reported a case of pressure wound in the margins of a removable maxillary denture in 66 year old female on oral alendronate sodium treatment for eight years.

DISCUSSION AND CONCLUDING REMARKS

Physicians and dentists alike must become increasingly aware of impaired oral healing following the use of BPs given for malignancy-related osteopenias,⁸⁶ and more effective communication process between prescribing physicians, dentists and patients using BPs, is needed.⁴⁹ Based on a case report of post-surgical complications, Soileau⁸⁶ suggested that oral/ dental examination should be performed before commencement of BP therapy as recommended for radiation therapy. To reduce any surgical complications after the instigation of BP therapy, precautionary treatment inclusive of periodontal, prosthetic and endodontic therapy, combined with appropriate dental extractions should be instituted.²⁹ Dental recall examinations should be performed throughout the course of BP therapy and once the therapy has been initiated, patients should be recalled every six months to ensure optimal oral health. Additionally, most of the times, this drug-induced osteonecrosis of jaw (BON), is detected not by a medical oncologist, but rather upon presentation for dental evaluation.⁸⁶ Published data revealed increasing reports of osteonecrosis associated with the nBPs, and oral surgeons most frequently have seen and treated such cases, which if remains untreated can lead complications like pansinusitis in untreated maxillary osteonecrosis.⁸⁵ Bone remodelling is severely affected in patients taking long term bisphosphonates therapy, therefore such patients are poor candidates for bone grafting, sinus lift and ridge augmentation procedure. Hence, American Association of Oral and Maxillofacial Surgeons (AAOMS)⁴⁷ suggested that patients be adequately informed of the potential risk of compromised bone healing and the risk of developing BON (Table 2), and

Table 2: American Association of Oral and Maxillofacial Surgeons (AAOMS) staging system for BON.⁴⁷

Stage 0	No clinical evidence of necrotic bone, but non-specific clinical findings and symptoms
Stage 1	Exposed/ necrotic bone in asymptomatic patients without evidence of infection
Stage 2	Exposed / necrotic bone associated with infection as evidenced by pain and erythema in region of exposed bone with or without purulent discharge.
Stage 3	Exposed/ necrotic bone in patients with pain, infection, and one or more of the following: exposed and necrotic bone extending beyond the region of alveolar bone resulting in pathological fracture, extraoral fistula, oral antral/ oral nasal communication, or osteolysis extending to the inferior border of the mandible or the sinus floor.

clinicians should follow-up patients carefully until surgical wounds are healed completely.⁸⁵ Although, the incidence of BON is much lower in patients receiving oral BPs therapy than patients receiving intravenous BPs therapy. However, such patients often requiring routine dental care should be treated cautiously, as oral BPs use has been found associated with a small risk of developing oral osteonecrosis that occur spontaneously or after the patient has undergone dental surgery.⁸⁵ Any established BON patient needing surgical procedures should be referred to an oral and maxillofacial surgeon, who may consult other qualified specialists as appropriate. The role of hyperbaric oxygen [HBO] and vascularized tissue transfers in the reconstructive management of bisphosphonate-related osteonecrosis of the jaw have yet to be revealed in BON cases.¹

In addition to radiographic imaging, a complete blood count may help in assessing the state of the patient in terms of possible infection. Assays to monitor bone turnover, such as N-telopeptide (NTx) and C-telopeptide (CTx) levels, which are reduced by BPs may help in future risk and diagnosis of BON.⁸⁷ Suppression of bone turnover can be an indicator for increased bone mineral density (BMD) and can possibly be related to its effects from bisphosphonates.⁸⁸

For a patient who has been taking an oral BP longer than three years, it should be discontinued, 3 months before and 3 months after the surgical procedure, if approved by the patient's physician. Serum CTx should be greater than 150 pg/mL before any surgical procedure, and rechecked at the time of surgery.⁸⁷ Non-surgical periodontal therapy is preferred. Any dentoalveolar surgical procedure (i.e. extractions, implants or apical surgery) should be avoided since the surgical sites will likely result in additional areas of exposed necrotic bone,

however, loose teeth, loose sequestra segments, but without exposing uninvolved bone. Similarly, to avoid trauma to the adjacent soft tissues, sharp bony edges should be removed. Systemic antibiotic therapy as well as chlorhexidine rinse, immediately before and after any inevitable surgical procedure for perioperative prophylaxis,⁸⁹ while establishing and maintaining meticulous oral hygiene, is essential. Furthermore, it should always be kept in mind, that knowledge base for bisphosphonate-associated BON is rapidly increasing, and it is likely that the current recommendations made by various societies^{88,90,91} may change over time.

Thus it can be concluded that;

- Physicians and dentists should be fully updated regarding the potential complications in the management of patients on BPs.
- Effective communication process between prescribing physicians, dentists and patients on BPs, is needed.
- BON is much lower in patients receiving oral BPs as compared with patients receiving intravenous BP therapy and good oral hygiene, accompanied by regular dental care using non-invasive procedures, is the best way to minimise this risk, if it exists.

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